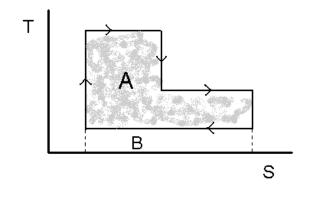
## Homework 2 (Due Thursday, February 2nd)

1. An engine is represented by the cyclic transformation shown in the T - S diagram below. The variable A denotes the area of the shaded region and B the area of the region below it. Find the efficiency in terms of the ratio of the areas A and B; show that this engine is not as efficient as a Carnot engine operating between the highest and lowest available temperatures.



2. Systems  $\mathcal{A}$  and  $\mathcal{A}'$  are isolated from the rest of the world but are in thermal and mechanical contact with each other (i.e., equal temperature and pressure). The total energy,  $U_{\Sigma} = U + U'$ , and total volume  $V_{\Sigma} = V + V'$  are constant. The entropy for system  $\mathcal{A}$  is  $S(U, V) = k \ln(CU^a V^b)$  Similarly, for system  $\mathcal{A}', S'(U', V') = k \ln(C'U'^c V'^d)$  where a, b, c, d, C, and C' are constants. Find U and V, the energy and volume of system  $\mathcal{A}$  at thermodynamic equilibrium in terms of  $U_{\Sigma}, V_{\Sigma}$  and the constants a, b, c, d.